

**W. E. Bachmann and L. H. Klemm.** Preparation and Reactions of 1-Cyclopentynaphthalene.

Page 4911. In col. 2, text line 5 from the end, for "COOH<sub>3</sub>" read "COCH<sub>3</sub>."—LEROY H. KLEMM.

**Tyunosin Ukita, Shoshichi Nojima and Makoto Matsumoto.** The Antibacterial Properties of Compounds Containing the Tricarbonylmethane Group. IV. Syntheses of 3-Alkylated or 3-Acylated 4-Hydroxycoumarins and of a Related Pyrone.

Page 5144. In Table I, last line, for "6-Methyl-2-hydroxy-3-octanoylpyrone-4" read "6-Methyl-4-hydroxy-3-octanoylpyrone-2."

In col. 1, line 5 of the text, for "6-methyl-2-hydroxypyrynone-4" read "6-methyl-4-hydroxypyrynone-2."

In col. 2, line 4 from the end, for "2-hydroxypyrynone-4" read "4-hydroxypyrynone-2."

**John P. Lambooy.** The Synthesis of 6,7-Diethyl-9-(D-1<sup>1</sup>-ribityl)-isoalloxazine.

Page 5226. In col. 1, line 48, for "o-diethylaniline" read "o-diethylbenzene."—JOHN P. LAMBOOY.

**John G. Milligan and Charles A. Kraus.** Chemistry of the Tris-(triphenylgermany)-silyl Group.

Page 5299. In col. 2, in the *Anal.* line, for "C<sub>64</sub>H<sub>48</sub>Ge<sub>3</sub>OH" read "C<sub>64</sub>H<sub>48</sub>Ge<sub>3</sub>SiOH." In Eq. (6) for "(Ph<sub>3</sub>Ge)<sub>3</sub>GeSiNH<sub>2</sub>" read "(Ph<sub>3</sub>Ge)<sub>3</sub>SiNH<sub>2</sub>."

**M. Sulzbacher.** 4-Benzoyloxy-acetophenone and -phenylmethylcarbinol.

Page 5346. The author states: "It has been brought to our attention that the substance 4-benzoyloxyphenylmethylcarbinol claimed as a new compound has already been described by Simonoff, *THIS JOURNAL*, 69, 2073 (1947), who used a different method of synthesis."—M. SULZBACHER.

**Homer Adkins, Robert M. Ross and Dorothy C. Schroeder.** The Synthesis of N-(2-Benzoyl-4-oxazoloyl)-valine.

Page 5401. In col. 2, lines 3 and 7, for "(VI)" read "(II)."—ROBERT M. ROSS.

**Alfred Burger and William B. Bennet.** Amino Derivatives of 2,2-Diphenylcyclohexanone. I.

Page 5415. In Table I, Col. 7, tenth and eleventh lines, <sup>k</sup> should be on 92-92.5.—ALFRED BURGER.

**F. E. Brown, Ralph E. Menzel, Meredith M. Stewart and Philip A. Lefrancois.** Cuprous Oxide as a Catalyst. I. Preparation and Promotion by Metallic Oxides.

Page 5605. In col. 2, line 3, for "1.07:7.14:1.4" read "10.7:7.14:14."—F. E. BROWN.

**N. Howell Furman and W. Charles Cooper.** A Study of the Polarographic Behavior of Dropping Amalgam Electrodes.

Pages 5669 and 5671. Figs. 2, 4 for "Microamp." read "—Microamp."

Page 5671. In col. 1, footnote (18) for "1942" read "1932." Line 15, for "positive" read "negative." Col. 2, line 26, for "lary had" read "lary maximum had."

Page 5672. Equation (1) in bracket part for "∂t" read "γt." Col. 2, line 24, for "Weischedel<sup>28</sup>" read "Weischedel<sup>27</sup>"; line 37, for "d<sup>1</sup>/s" read "D<sup>1</sup>/s."

Page 5675. Fig. 9, for "i<sub>a</sub>/m<sup>2</sup>/t<sup>1/2</sup>" read "—i<sub>a</sub>/m<sup>2</sup>/t<sup>1/2</sup>," and for "10<sup>4</sup>" read "10<sup>6</sup>."—N. HOWELL FURMAN.

**W. F. Giauque, R. E. Barieau and J. E. Kunzler.** Crystal Perfection of ZnSO<sub>4</sub>·7H<sub>2</sub>O. Partial Molal Heat Capacity, Heat Content and Vapor Pressure of its Aqueous Solutions. Thermodynamics of Clark Cell.

Pages 5685-5690. The two values of e.m.f. given for 313.16°K. in Table VIII should be interchanged. αF<sub>(44)</sub> in Table VIII should be αF<sub>(37)</sub>.—W. F. GIAUQUE.

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**Edward G. Taylor, Robert P. Desch and Arthur J. Catotti.** The Conductance of Sulfamic Acid and Some Sulfamates in Water at 25° and Conductance Measurements of Some Long Chain Sulfamates in Water and in Water-Acetone Mixtures at 25°.

Page 75. In cols. 1 and 3 of Tables I, II and III the heads should be 10<sup>2</sup>√C and not 10√C.

Page 76. In Table IV, for sulfamic acid read "398.31" instead of "393.31." In Figs. 3 and 4 the numbers reading along the abscissa should be 0, 2, 4, 6, and 8 instead of 2, 4, 6, and 8.

Page 78. In Col. 1, line 3, the word "must" should be between the words "sulfamates" and "lie."—EDWARD G. TAYLOR.

**Scott Searles.** The Reaction of Trimethylene Oxide with Grignard Reagents and Organolithium Compounds.

Page 125. In Table I, line 2 of the last col., for "96-96.3" read "86-86.3."—SCOTT SEARLES.

**Louis Meites and Thelma Meites.** Some Polarographic Effects of Gelatin and Other Maximum Suppressors.

Page 177. In col. 2, line 3, for "0" read "σ."

Page 178. In Fig. 3, the ordinate legends of Parts I and II should be transposed.

Page 181. In col. 1, line 13, for "—1.6" read "—1.4."—LOUIS MEITES.

**Floyd J. Thaller, Donald E. Trucker and Ernest I. Becker.** The Absorption Spectra of Some Chloro-substituted Tetracyclones.

Page 228. In Table I, m.p. col., transpose the 2nd and 3rd, and the 6th and 7th sets of values.—ERNEST I. BECKER.

**Sidney W. Fox and Yutaka Kobayashi.** Studies on Antipodes. VII. Observations on D-Amino Acids and on Casein Derivatives Containing D-Amino Acid Residues.

Page 354. In col. 2, line 5 in Experimental, for "+30.1 ± 0.5°" read "—30.1 ± 0.5°."—YUTAKA KOBAYASHI.

**Cyrus O. Guss.** Intramolecular Displacement of Carboxylate Ion. II. The Effect of Some Variations in Structure of Phenol-alcohols.

Page 609. In col. 1, line 16, after "phenol," read "anion, a property enhanced by a decrease in acid strength of the phenol."—CYRUS O. GUSS.

**John D. Roberts, Donald R. Smith and C. C. Lee.** The Decarbonylation of Diphenyl Triketone.

Page 622. In col. 2, the equation near the top is a part of footnote (21).—JOHN D. ROBERTS.

**Stanley J. Cristol, Norman L. Hause and John S. Meek.** Mechanisms of Elimination Reactions. III. The Kinetics of the Alkaline Dehydrochlorination of the Benzene Hexachloride Isomers. II.

Page 675. In col. 1, line 25, for "2.303/(b - a)" read "2.303/(b - 3a)."—STANLEY J. CRISTOL.